

Use of SNPs with Controlled Size & Shape for Enhanced Surface Hydrophobicity & Hardness for Coil Coating Applications

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OUTLINE

- Introduction
- Materials & Method
- Characterization & Testing
- Findings
- Conclusion
- References







INTRODUCTION



Organic coatings are used mainly for protection and decoration

Coatings used in high speed lines to coat coils (metals shaped into foils and winded as a roll)









INTRODUCTION

- **Substrate:** HDG or Aluminium

- Coating Line Speed: 100 m/min (curing at 240 °C in 20-30 seconds)

- **Dry Film Thickness:** 5 micron primer + 20 micron topcoat (in general)

Application Areas of Coil Coatings







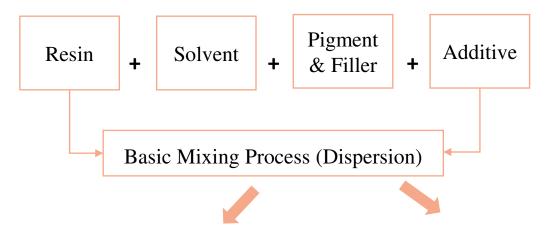








INTRODUCTION



Mechanical control



Fig 1. Impact Resistance Test



Fig 2. MEK Test



Color control

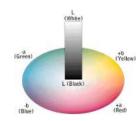


Fig 3. Color Control Test





Hydrophobic surfaces has good water repellency and provide excellent properties such as:

Decreasing corrosion rate, easy to clean, self-cleaning, self healing properties etc.

AIM OF THE STUDY

Synthesis of nano-inorganic particles to use in enhancing the hydrophobicity & hardness of coil coating surface.

HOW?

Creating roughness on the surface







Wetting

- It is strong indicator for describing the interfacial relationships btw L & S & G.
- Liquid molecules interact more strongly with solid surface than the liquid.

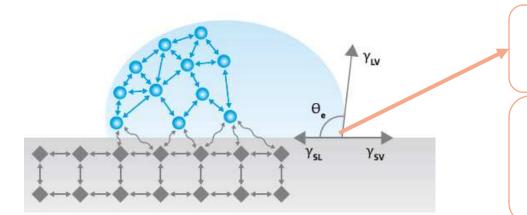


Fig 4. Schematic diagram of interfacial tension in vapor/liquid/solid system (Wang et al., 2020)

Contact Angle

Interfacial tensions form the equilibrium contact angle of wetting.

Young's Equation

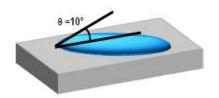
Assumes surface is ideal (flat, smooth, chemically homogenous)

$$\gamma_{sv} = \gamma_{sL} + \gamma_{Lv} \cos \Theta_Y$$

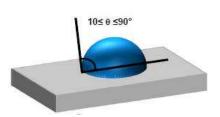






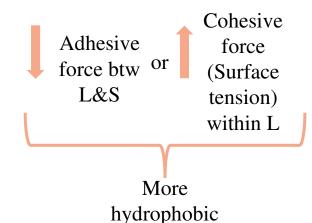


good wettability (hydrophilic)

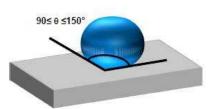


"normal" wettability

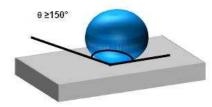






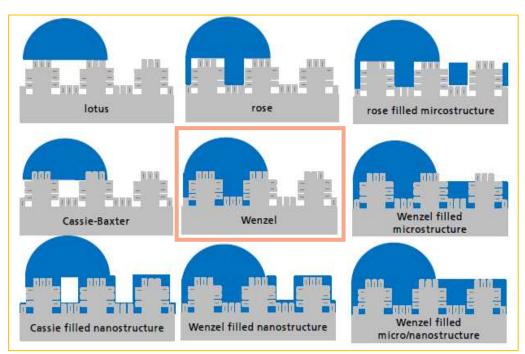


poor wettability (hydrophobic)

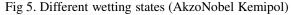


extremely poor wettability (super-hydrophobic)





- There can be different states of water droplet on a rough surface.
- Surface roughness and morphology lead enhancement of hydrophobic behavior of the material.















Classie's state



Transitional state btw Wenzel and Classie



"Lotus"

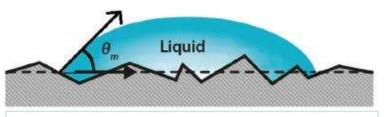
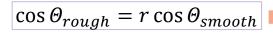


Fig 6. Apparent or measured contact angle on rough (Wenzel) surface (AkzoNobel Kemipol)







 $\cos \Theta_m = r \cos \Theta_Y$



r=1 for smooth surface,r>1 for rough surface



 $r = \frac{Actual\ surface\ area}{Planar\ surface\ area}$





MATERIALS & METHOD

Synthesis & Characterization of SNP

- Synthesis of silica by sol-gel
 / Stöber method and TEOS
- Characterization of SNP by SEM

Dispersion of SNP in coil coating

- Addition of NSP to PE coating mixture as a filler and dispersing.
- Characterization of coating mixture (Rheology, PSD)

Characterization of Coating

- Curing the coating for 40 sec at 240 °C and quenching
- Mechanical tests
- Surface characterization by CA & Surface energy measurements, SEM, AFM



Both monosize and multisize silica particles were added to coating mixture separately.







CHARACTERIZATION & TESTING

Monosize silica particles

Synthesis of Stöber Silica Particles

***** Effect of TEOS Concentration

Bi-modal silica particles (Seed Addition / Seeded Growth)

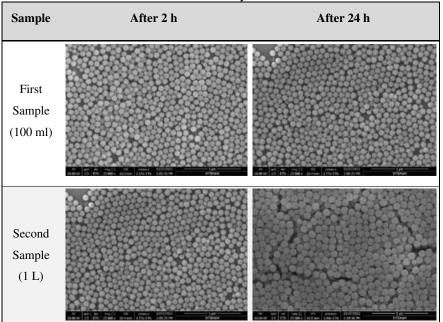






CHARACTERIZATION & TESTING(Synthesis of Stöber Silica Particles)

Table 1. SEM images of silica particles synthesized with different batches after 2 h and 24 h of synthesis



- ❖ Perfect spherical SNPs with 500-550 nm
- Repeatability & reproducibility for small & large scale samples
- Obtaining same particles after
 2h and 24 h at the beginning of reaction







CHARACTERIZATION & TESTING

Monosize silica particles

* Synthesis of Stöber Silica Particles

***** Effect of TEOS Concentration

Bi-modal silica particles (Seed Addition / Seeded Growth)



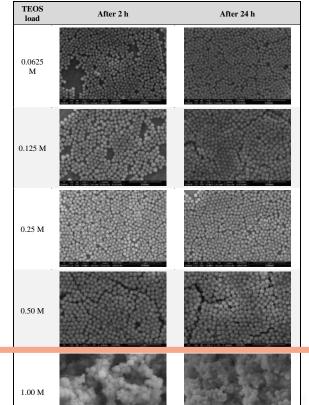




CHARACTERIZATION & TESTING (Effect of TEOS Concentration)

Table 2. Effect of TEOS concentrations

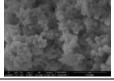
| Parameters | Value |
|--------------------------|---|
| TEOS (mol/L) | 0.0625M, 0.125M, 0.25 M , 0.50 M, 1 M |
| Ethanol (mol/L) | 12.14 M |
| H ₂ O (mol/L) | 11.67 M |
| NH ₃ (mol/L) | 1.09 M |











CHARACTERIZATION & TESTING(Effect of TEOS Concentration)

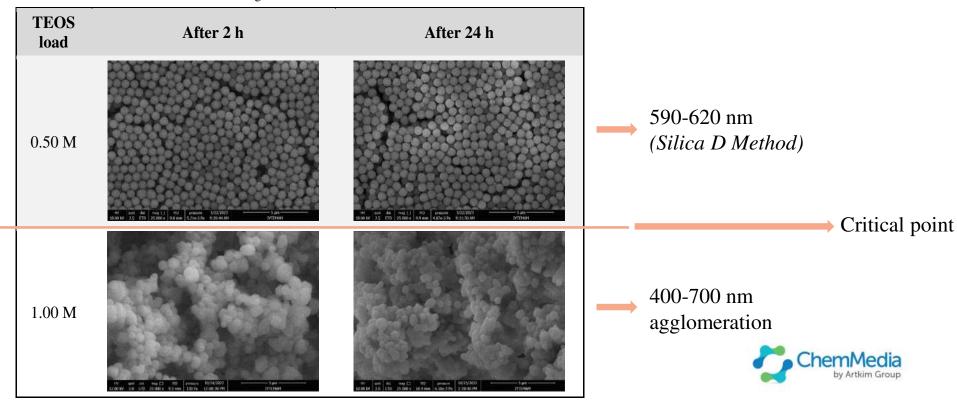
Table 3. SEM images of effect of TEOS concentrations

| TEOS load | After 2 h | After 24 h |
|--------------|--|---|
| 0.0625 M | THE REST OF THE PARTY OF THE PA | THE RESIDENCE OF THE PROPERTY |

400-470 nm

CHARACTERIZATION & TESTING(Effect of TEOS Concentration)

Table 4. SEM images of effect of TEOS concentrations



CHARACTERIZATION & TESTING

Monosize silica particles

* Synthesis of Stöber Silica Particles

***** Effect of TEOS Concentration

Bi-modal silica particles (Seed Addition / Seeded Growth)







CHARACTERIZATION & TESTING

(Seed Addition / Seeded Growth)

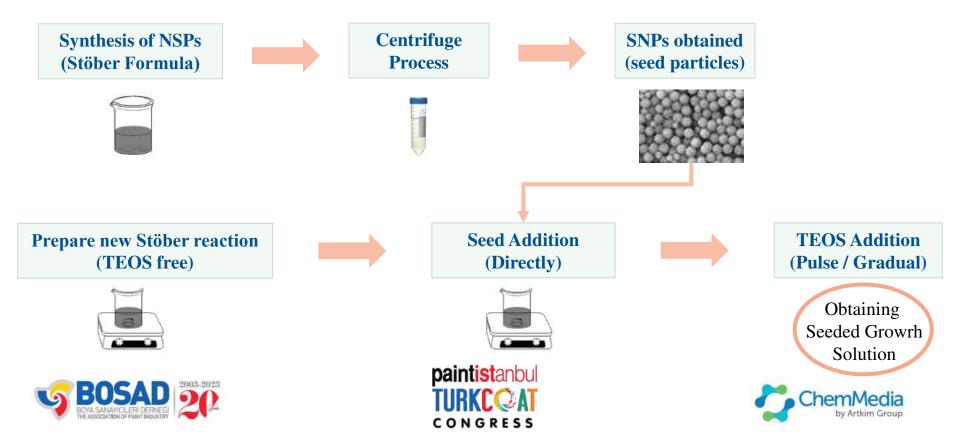
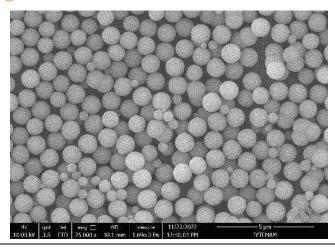


Table 5. Seed: Classic Stöber SNPs and their different growth solutions

| Table 3. Seed. Classic Stood Styl 5 and their different growth solutions | | | | |
|--|-----------|--|--|--|
| Method | After 2 h | Method | After 2 h | |
| Seed: Stöber Method (Silica C) (500-550 nm) | (4) | 2nd batch - Silica D Method (Obtained particles: 1.1μm + 200 nm) | 77 141 142 143 1 | |
| 1st batch – Classic Stöber Method with doubled TEOS and water concentrations (Obtained particles: 900 nm + 500-600 nm) | ** | 3rd batch - Silica D method with doubled TEOS and EtOH concentrations (Obtained particles: 650 – 750 nm) | 100 to 7 15 17 17 2500 to 52 20 10 10 10 10 10 10 10 10 10 10 10 10 10 | |

Seed: Classic Stöber SNPs & growth solution with Silica D method

2nd batch - Silica D Method (Obtained particles: 1.1µm + 200 nm)



Particles with 200 nm smaller than the initial seed size. This indicates that new reactions were still ongoing, and a gradual addition of TEOS was necessary instead of a pulse addition.

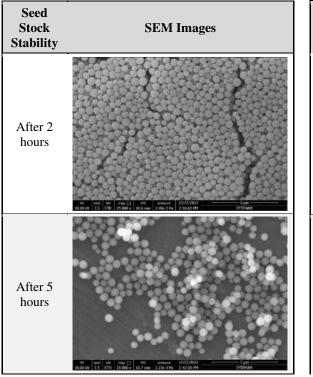


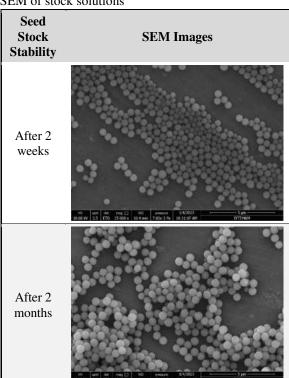




Preparing stock solution of SNPs

Table 6. SEM of stock solutions





- Producing Stöber SNPs
- Preparing stock solution with a solid concentration of 15 g/L
- Short and long term stability



Table 7. SEM of growth solutions (75 mg seed)

| | Tuble 7. BENT of growth solutions (75 mg seed) | | | |
|--|---|---|--|--|
| TEOS &EtOH | After 4 h | After 24 h | | |
| 1 ml/min (Obtained particles: 1μ + 200 nm) | Wi onl Se Fig. 260 prints 12200022 | W ₁ cod M1 'N6 ⊥ 100 '9788N" 123V9507 - 3 pil- 12304V 3.5 CTD 12506.2 163 ers 1260-176 123040 M1 27704M | | |
| 8 ml/min (Obtained particles: 1.2µ + 200 nm) | 70 pt 1 00 80 700 count (107/07) 100 100 100 100 100 100 100 100 100 10 | up set cont. 65 cont. 17/7/207 | | |

TEOS/EtOH ratio: 1/10

Addition rate: 1 & 8 ml/min

Seed amount: 75 mg Stöber SNPs

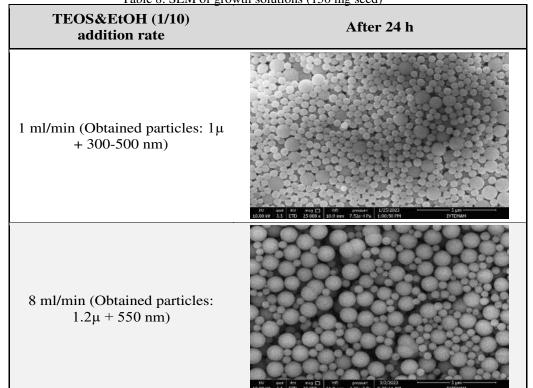
TEOS molecules started generating smaller nanoparticles around 500 nm instead of diffusing through the seed particles for growth. Therefore, overnight duration was required to complete the reaction and diffusion processes.







Table 8. SEM of growth solutions (150 mg seed)



TEOS/EtOH ratio: 1/10

Addition rate: 1 & 8 ml/min

Seed amount: 150 mg Stöber SNPs

Rate of 8 ml/min: more equal size

distribution



Table 9. SEM of 100 ml & 1L growth solutions (150 mg seed)

| TEOS&EtOH (1/10) addition rate | After 24 h | |
|--|---|--|
| 8 ml/min 100 ml solution (Obtained particles: 1.2μ + 550 nm) | 100 0W 33 CTO 25000 x 110 ores 4.58-3 Pt 0.58-14 M | |
| 8 ml/min 1.5 L solution (Obtained particles: 1.2μ + 550 nm) | 17 194 56 109 7 97 PRAST 3/2/3923 5 5 5 5 5 5 5 5 5 | |

TEOS/EtOH ratio: 1/10

Addition rate: 8 ml/min

Seed amount: 150 mg Stöber SNPs

Both 100 ml and 1.5 L solutions to control reproducibility



CHARACTERIZATION & TESTING (Coating Studies)

Monosilica Addition to Topcoat

Bi-modal silica Addition to Topcoat

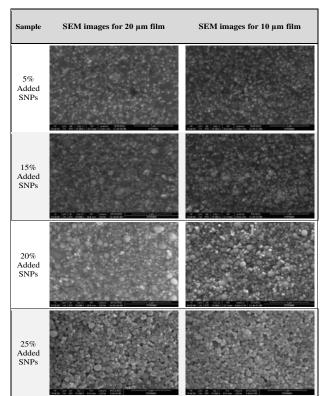


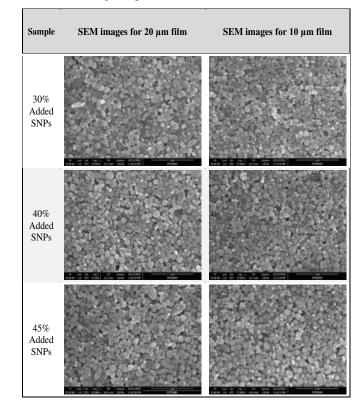




CHARACTERIZATION & TESTING(Coating Studies – Monosilica Addition)

Table 10. SEM of monosilica added coating samples





similar images with 25% added SNPs



CHARACTERIZATION & TESTING(Coating Studies – Monosilica Addition)

Contact Angle Measurements

Table 11. CA for 20 & 10 μm films

| Loading | | Monosize - 10 mikron |
|---------|--------|-------------------------|
| 45% | 113.61 | 111.90 |
| 40% | 113.54 | 110.80 |
| 35% | 112.00 | 109.50 |
| 30% | 110.50 | 105.42 |
| 25% | 107.60 | 107.34 |
| 20% | 92.00 | 92.71 |
| 15% | 87.66 | 87.25 |
| 10% | 86.32 | 84.2 |
| 5% | 85.45 | 83.50 |
| 0% | 82.20 | 82.20 |

| Table 12. Rought | iess Paramete | rs & Actual | Contact Angle |
|------------------|---------------|-------------|---------------|

| Sample | Projected Surface Area (μ/m²) | Textured Surface Area (μ/m²) | Roughness Parameter | Measured Contact Angle (°) | Actual Contact Angle (°) |
|--------|----------------------------------|---------------------------------|------------------------|----------------------------------|--------------------------------|
| 45% | 25.00 | 28.81 | 1.15 | 113.61 | 117.49 |
| 40% | 25.00 | 28.98 | 1.16 | 113.54 | 117.58 |
| 30% | 25.00 | 29.35 | 1.17 | 110.5 | 114.28 |
| 20% | 25.00 | 28.25 | 1.13 | 92.00 | 92.26 |
| 10% | 25.00 | 25.52 | 1.02 | 86.32 | 86.24 |

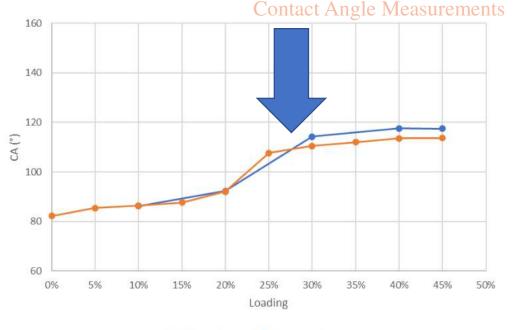
$$\cos \Theta_m = r \cos \Theta_Y$$



$$r = \frac{Actual\ surface\ area}{Planar\ surface\ area}$$



CHARACTERIZATION & TESTING(Coating Studies – Monosilica Addition)

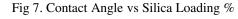


Obtaining 90° between 15-20 %

Not dramatically increasing after loading of 25%

Maximum obtained CA: 113.61° at 45%











CHARACTERIZATION & TESTING

(Coating Studies – Monosilica Addition)

Mechanical Tests

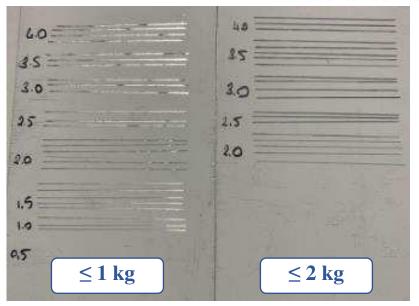


Fig 8. Scratch resistance results of 25 % and 30 % monosize silica added samples respectively



Table 13. Pencil hardness results of both 20 μm and 10 μm

| Sample | Results of 20 micron | Results of 10 micron |
|-----------|----------------------|----------------------|
| 0 - 35 % | 2Н | 2Н |
| 40 - 45 % | 3Н | 3Н |

Loading of Monosize SNPs



More hydrophobic & harder coating surface





CHARACTERIZATION & TESTING (Coating Studies)

Monosilica Addition to Topcoat

Bi-modal silica Addition to Topcoat







CHARACTERIZATION & TESTING (Coating Studies – Bi-modal Silica Addition)

Table 14. AFM & SEM results of bi-modal SNPs added coating samples

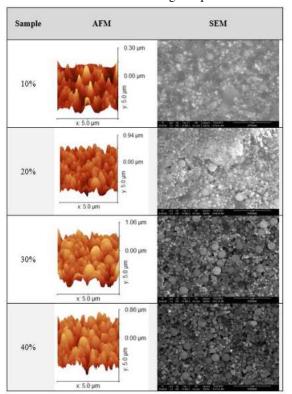


Table 15. CA for 20 & 10 μm films

| Load | Bi- modal 20 µm | Bi-modal 10 μm | |
|------|-----------------------|-------------------|---|
| 40% | 116.70 | 111.50 | |
| 35% | 104.30 | 114.30 | |
| 30% | 108.40 | 110.50 | - |
| 25% | 107.90 | 109.90 | |
| 20% | 106.10 | 106.40 | - |
| 15% | 87.90 | 90.40 | |
| 10% | 88.30 | 88.50 | |
| 5% | 88.20 | 89.70 | |
| 0% | 82.20 | 82.20 | |

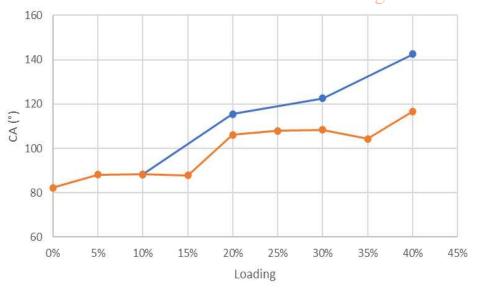
Table 16. Roughness Parameters & Actual Contact Angle

| | Sample | Projected Surface Area (μ/m²) | Textured Surface Area (µ/m²) | Roughness Parameter | Measured Contact Angle (°) | Actual Contact Angle (°) |
|---|--------|-------------------------------------|---------------------------------------|------------------------|----------------------------------|-----------------------------------|
| | 40% | 25.00 | 44.13 | 1.77 | 116.70 | 142.48 |
| | 30% | 25.00 | 42.64 | 1.71 | 108.40 | 122.57 |
| | 20% | 25.00 | 38.85 | 1.55 | 106.10 | 115.53 |
| | 10% | 25.00 | 25.96 | 1.04 | 88.30 | 88.24 |
| 1 | | | | | | |



CHARACTERIZATION & TESTING(Coating Studies – Bi-modal Silica Addition)

Contact Angle Measurements



Obtaining 90° between 10-15 %

Maximum obtained CA: 116.7° at 40%

Maximum CA: 142.48° by considering roughness parameter



Fig 9. Contact Angle vs Silica Loading %





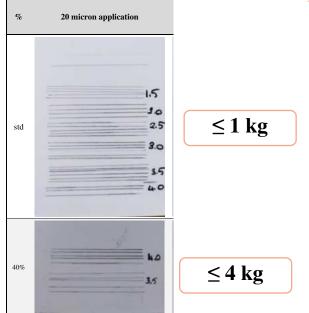


CHARACTERIZATION & TESTING(Coating Studies – Bi-modal Silica Addition)

Mechanical Tests

Table 17. Pencil hardness results of both 20 μm and 10 μm

| Sample | | Results of 10 micron |
|-----------|----|----------------------|
| 0 - 20 % | 2Н | 2Н |
| 20 - 40 % | 3Н | 3Н |



Loading of multisize SNPs

More hydrophobic & harder coating surface

Fig 10. Scratch resistance results of std and 40 % bimodal silica added samples for both 10 & 20 μ m.



Bi-modal silica particles have more effect on improving the scratch resistance of the surface than mono-size SNPs.

FINDINGS

Mono (550 nm) and bi-modal (550-1200 nm) nano silica particles were synthesized and employed as additives to increase the hardness and contact angle simultaneously in commercial coil coatings.







FINDINGS

The CA value of standard coat was 82°. It was increased to 113.61° with 45% monosized silica addition.

Monosized silica nanoparticles (35%) resulted in a surface hardness of 2H.

Increasing the loading to 45% improved the surface hardness to 3H.

40% addition of monosize silica was necessary to achieve a pencil hardness of 3H.







FINDINGS

Highest CA value achieved was 116.7° with addition of 40% bi-modal nano silica particles.

Incorporating the roughness parameter implies an effective CA value of 140°.

20% addition of bi-modal silica was sufficient to achieve a surface hardness of 3H.







CONCLUSION

Addition of silica nano particles simultaneously improves hardness and surface hydrophobicity in coil coating applications.

Bi-modal silica nano particles results in better contact angle and surface hardness performance compared to mono-modal particle size distribution.







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CONGRESS

